

■ V.V. Titanium Pigments Pvt. Ltd. ■

(AN ISO 9001 : 2008 | ISO 14001 : 2004 | OHSAS 18001 : 2007 CERTIFIED COMPANY)

Rev Date: 08.02.2016

Revision: 03

## GENERIC MATERIAL SAFETY DATA SHEET (MSDS) FOR FERROUS SULPHATE HEPTAHYDRATE

NO	SECTIONS	DETAILED EXPLANATION
1	Identification of the substance	Other names: Iron (II) Sulphate Heptahydrate, Iron Vitrol; Green Vitrol; Ferrous sulphate. U.N.No: None Allocated. DG Class: None Allocated. Sub risk: None Allocated. Hazchem: None Allocated. Packing group: None Allocated. Posion schedule: None Allocated. Uses: Water and sewage treatment; reducing agent, wood preservative; fertilizer.
2	HAZARD IDENTIFICATION	No adverse health effects expected if the product is handled in accordance with this Safety Data Sheet and the product label.
3	COMPOSITION / INFORMATION ON INGREDENTS	Chemical Name CAS Number Proportion Ferrous Sulphate Heptahydrate 7782-63-0 >98.0%
4.	FIRST AID MEASURES	Swallowed: If conscious wash out mouth with water and give water to drink. Do not induce vomiting. Seek immediate medical attention. Eye: Immediately irrigate with copious quantities of water for at least 15 minutes. Eyelids to be held open. Seek medical advice. Skin: Wash with plenty of soap and water. Remove contaminated clothing, launder clothes before reuse. If irritation persists seek medical advice. Inhaled: Remove victim from exposure to fresh air immediately. If not breathing apply artificial respiration. If breathing is difficult given oxygen. Seek immediate medical advice.
5	FIRE FIGHTING MEASURES	Non combustible product. Use extinguishing media appropriate for surrounding fire. When heated to decomposition may produce toxic SOX fumes. Fire fighters to wear self – contained breathing apparatus if risk of exposure to vapour or products of combustion.
6.	ACCIDENTAL RELEASE MEASURES	Wear protective equipment to prevent skin and eye contamination and inhalation of dust. Sweep up but avoid generating dust. Collect and place in suitably labeled containers.
7.	HANDLING AND STORAGE	Not defined as a dangerous good by the Australian Code for the Transport of Dangerous Goods by Sea, Road, Rail & Air.
8.	EXPOSURE CONTROLS AND PERSONAL PROTECTION	No value assigned for this specific material by the national occupational health and safety commission (work Australia). Exposure standard for iron salts as fe:  TWA: 1 mg/m3  Engineering controls: Use adequate general or local exhaust ventilation to keep exposure levels below the recommended exposure standard. Personal protection: Avoid skin and eye contact and inhalation of dust. Wear overalls, eye protection and gloves. Use chemical safety goggles or face shield where dusting or splashing of solutions is possible. Use a dust mask or dust/mist respirator in dusty conditions. Always was hands before smoking, eating, drinking or using the toilet.

Factory: A-81, SIPCOT Industrial Complex, South Veerapandiapuram PO, Thoothukkudi - 628002. Tamil Wedy TNDIA ORIN

Phone: +91-461-2340491 / 92 / 93 / 94 / 95. Fax: +91-461-2340449 Email: info@vvtipigments.com web: www.vtipigments.com PAN: AADCV7723P CIN: U24110TN2011PTC082322

Registered Office: Keeraikaranthattu, Tisaiyanvilai - 627657. Tirunelveli District. Tamil Nadu, INDIA. Ph: +91-4637-272361. Fax: +91-4637-272802.







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		Substances (11100)
16.	PRINCIPAL REFERENCES	This chemical is listed on the Australian inventory of chemical substances (AICS)
9		TSCA 8(b) inventory: All components listed. TSCA 12(b) annual export notification: lead.
	9	U.S. Federal regulations: TSCA 6 proposed risk management: Lead
		Irritating material Target organ effects
	INFORMATION	HCS Classification : Toxic material
15.	REGULATORY	United States
	INFORMATION	Transport of Dangerous Goods by Sea, Road, Rail and Air.
14.	TRANSPORT	Not defined as a dangerous good by the Australian Code for the
13.	DISPOSAL CONSIDERATIONS	
10	DICDOCAL	not breathe dust.  Refer to Stage land Waste Management Authority.
		SAFETY PHRASES: S24/25 Avoid contact with eyes and skin, S22 do
		RISK PHRASES: R22 HARMFUL IF SWALLOWED.
12.	INFORMATION	Hazard Category: Harmful
12.	ECOLOGICAL	the spleen and liver.  No data available.
		the liver, spleen and Lymphatic system. This could lead to damage to
		CHRONIC EFFECTS: Repeated exposure may increase iron levels in
		tract.
		INHALED: Inhalation of dust may cause irritation to the respiratory
	8	SKIN: May cause irritation to skin.
		EYE: May cause mild eye irritation.
		Evidence from animal tests indicate that repeated or prolonged exposure to this chemical could result in damage to liver and kidneys.
		cyanosis, pulmonary oedema and convulsions leading to coma.
	INFORMATION	diarrhea and abdominal pain. Ingestion of large doses can lead to
11.	TOXICOLOGICAL	Harmful if swallowed. Ingestion can result in nausea, vomiting,
		Substance oxidizes on contact with moisture to form ferric sulphate.
	Sold Street, April 2007 18 10 2	potassium iodide, silver salts, sodium borate, sodium tartrate tannin.
	REACTIVITY	alkalis, carbonates, gold salts, lead acetate, lime water, potassium,
10.	STABILITY AND	Stable under normal temperatures and pressures. Incompatible with
		Molecular formula : FeSO4.7 H20
		pH (10% solution) : 3.7 Solubility in water : 48.6 g/100g @ 50 Deg C
		Flash Point (closed cup): Not Available
		Vapour Density (Air=1): Not Available
		Specific Gravity (20 Deg C) :1.89
		Boiling Point: 300 Deg C
J.	CHEMICAL PROPERTIES	Melting Point : 64 Deg C
9.	PHYSICAL AND	Appearance, odour: Blue-green crystalline solid, odourless.

